

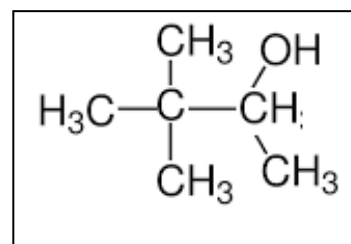
Name

51

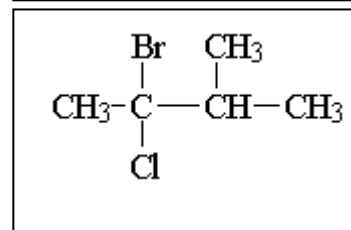
Year 11 Organic Chemistry test

1) Name the following compounds

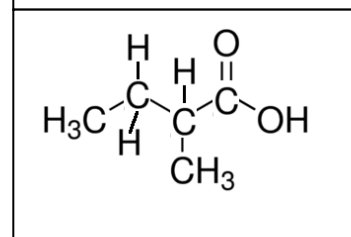
a) _____



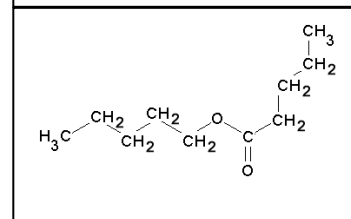
b) _____



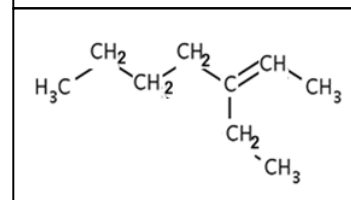
c) _____



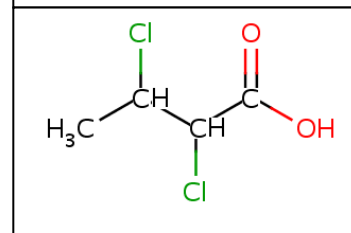
d) _____



e) _____



f) _____



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2) Draw structural formulae of the following compounds

a) 2,4-dimethylpentane



b) 2-bromo-3-chloropent-1-ene



c) 4-methylpent-2-yne



d) Butyl ethanoate



e) 3-chloropent-2,3-diol



f) 2-bromobutanoic acid



3) Name and draw the structural formula for two isomers with the molecular formula C_4H_{10}

Name _____

Name _____

4 marks

4) A 400.0 gram sample of hexane is obtained. What mass of the sample is due to carbon?
Show all working out.

2 marks

5) Name all the possible products from an addition reaction between:

a) Propene and Cl_2 gas.

b) But-2-ene and HCl gas

2 marks

6) Write a balanced chemical equation for the combustion of butane gas.

3 marks

7) What reactants must be mixed to form butyl propanoate.

2 marks

- 8) Place the following in order of increasing solubility in water.
Hexanoic acid, methanoic acid, ethanoic acid, butanoic acid

highest

↑

lowest

4 marks

- 9) Place the following in order of increasing melting temperature.
Hexan-1-ol, methanol, ethanol, butan-1-ol

highest

↑

lowest

4 marks

- 10) Complete the table below.

Name	Structural formula	Semi-structural formula
	$\begin{array}{ccccccc} \text{CH}_3 & - & \text{CH}_2 & - & \text{CH} & - & \text{CH}_2 & - & \text{CH}_3 \\ & & & & & & & & \\ & & & & \text{CH}_3 & & & & \end{array}$	
1,3-dibromo-4-methylhexane		
		$\text{CHCl}_2\text{CH}_2\text{C}(\text{CH}_3)_2\text{CH}_2\text{CHCH}_2$

6 marks