

- 1. Consider the pathways shown above. Using these pathways answer the following questions.
 - a. Name the following compounds
 - A ______
 - B _____
 - E _____
 - X _____
 - G
 - D _____
 - L
 - b. Draw the structural formula of X.
 - i. to what class of compounds does F belong to? ______
 - ii. to what class of compounds does G belong to?
 - c. Identify the reagents below:
 - i. "j". iii "h".
 - d. Identify "k" by placing a circle around the most likely option.

 $MnO_4^-(aq)/H^+(aq)$, $H_2O(g)/H_3PO_4(s)$, HCI(aq), NaOH(aq)

e. Identify the type of reaction that forms:

i. E ______ ii. J ______

iii. K

- f. Consider the formation of substance F.
 - i. Give the oxidation and reduction half equations for the reactions taking place in the formation of substance F given that $Cr_27_4^{2-}$ (aq) \rightarrow Cr^{3+} (aq)
 - ii. Give the balanced overall equation, states included.